

ammonia (NH3), which is a colorless gas with a strong, pungent odor. It is highly soluble in water, forming ammonium hydroxide. Ammonia is commonly used in agriculture as a fertilizer and in industry for various purposes. It is also a key component in the production of many chemicals. The boiling point of ammonia is -33 °C, and it has a molecular weight of 17.03. Ammonia is produced by the Haber-Bosch process, which involves the reaction of nitrogen and hydrogen under high pressure and temperature. It is a colorless, odorless gas at room temperature, but it has a sharp, irritating smell. Ammonia is used in a wide range of applications, from household cleaning to industrial processes. It is also a natural product of many living organisms. The chemical structure of ammonia is a central nitrogen atom bonded to three hydrogen atoms. It is a weak base and can act as a catalyst in many reactions. Ammonia is also used in the production of explosives and as a refrigerant. It is a colorless, odorless gas at room temperature, but it has a sharp, irritating smell. Ammonia is used in a wide range of applications, from household cleaning to industrial processes. It is also a natural product of many living organisms. The chemical structure of ammonia is a central nitrogen atom bonded to three hydrogen atoms. It is a weak base and can act as a catalyst in many reactions. Ammonia is also used in the production of explosives and as a refrigerant.

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